

How to Minimize Problems with Electrode Passivation in Electroanalytical Chemistry

Jiri Barek

Head of the UNESCO Laboratory of Environmental Electrochemistry, Charles University, Faculty of Science,
Department of Analytical Chemistry, Albertov 6, 128 43 Prague 2, Czech Republic
Email: barek@natur.cuni.cz

ABSTRACT

Passivation of electrode materials is one of the biggest problems in practical applications of modern electroanalytical methods in analytical laboratories. Approaches used in our UNESCO Laboratory of Environmental Electrochemistry to minimize this problem will be discussed.

Keywords: Amperometry; Electrode passivation; Novel electrode materials; Voltammetry.

INTRODUCTION

Approaches to minimize working electrode passivation can be briefly classified as follows [1]

- To use electrodes with renewable surface
- To use disposable electrodes
- To use surface modification preventing passivation
- To use measurements in flowing systems minimizing passivation
- To prevent access of passivating compounds to electrode surface
- To use novel electrode materials resistant to passivation
- To use heated electrodes
- To use enzymes converting passivating compounds to non-passivating
- To use intentional electrode fouling

OBJECTIVES

In spite of a number of papers dealing with electrode passivation, non-systematic and more or less random approach still prevails in this field. Therefore, it is desirable to investigate the relationship between structure of the electrode surface, structure of passivating substance and electrochemical behaviour of passivating substances to clarify nature and mechanism of these processes and to find ways how to diminish, suppress or completely eliminate passivation and connected negative effects complicating electroanalysis of real samples.

METHODOLOGY

We have developed or applied many suitable measuring protocols compatible with minimizing electrode passivation. See the following examples

- Silver amalgam electrodes with renewable surface [2]
- Boron doped diamond electrodes resistant to passivation [3]
- Batch or flow injection analysis to suppress electrode passivation [4]
- To use hollow fibers to prevent access of passivation compounds to electrode surface [5]
- To use enzymes to circumvent electrode passivation [6]

CONCLUSION/RESULTS

We believe that most promising approach from the point of view of practical analytical laboratories is the search for novel electrode materials more resistant to passivation and their combination with measurement in flowing systems. Other approaches are also efficient but slightly more complex and thus not so attractive to practical laboratories.

ACKNOWLEDGMENT

The authors would like to express appreciation for the support of the Grant Agency of the Czech Republic (project 20-01589S). We appreciate material, technical and intellectual support of Metrohm.CZ (<https://www.metrohm.com/cs-cz/>).

REFERENCES

1. Barek, Jiri." How to Improve the Performance of Electrochemical Sensors via Minimization of Electrode Passivation". CHEMOSENSORS, Volume 9, Issue 1, Article Number 12. DOI 10.3390/chemosensors9010012."
2. Gajdar, Julius; Stafurova, Kristina; Barek, Jiri; Fischer, Jan. „Retractable-pen-based renewable silver amalgam film electrode for microliter sample analysis of electrochemically reducible environmental pollutants". SENSORS AND ACTUATORS B-CHEMICAL, Volume 329, Article Number 129057, DOI 10.1016/j.snb.2020.129057.
3. Schwarzova-Peckova, Karolina; Vosahlova, Jana; Barek, Jiri; Sloufova, Ivana; Pavlova, Ewa; Petrak, Vaclav; Zavazalova, Jana." Influence of boron content on the morphological, spectral, and electroanalytical characteristics of anodically oxidized boron-doped diamond electrodes". ELECTROCHIMICA ACTA, Volume 243, page 170-182, year 2017. DOI 10.1016/j.electacta.2017.05.006.
4. Jiranek, Ivan; Barek, Jiri" The use of non-traditional carbon film electrode based on microcrystalline natural graphite - polystyrene composite film for amperometric determination of 5-aminoquinoline using flow injection analysis minimising electrode fouling. ". JOURNAL OF ELECTROANALYTICAL CHEMISTRY, Volume 885, Article Number 115085, DOI 10.1016/j.jelechem.2021.115085.
5. Hrdlicka, Vojtech; Barek, Jiri; Navratil, Tomas. "Differential pulse voltammetric determination of homovanillic acid as a tumor biomarker in human urine after hollow fiber-based liquid-phase microextraction". TALANTA Volume 221, Article Number 121594, DOI 10.1016/j.talanta.2020.121594.
6. TvorynskaSofiiia, Barek Jiri, Josypcuk Bohdan. "Influence of different covalent immobilization protocols on electroanalytical performance of laccase-based biosensor". Biosensors, Volume 148, Article number 108223, DOI10.1016/j.bioelechem.2022.108223.